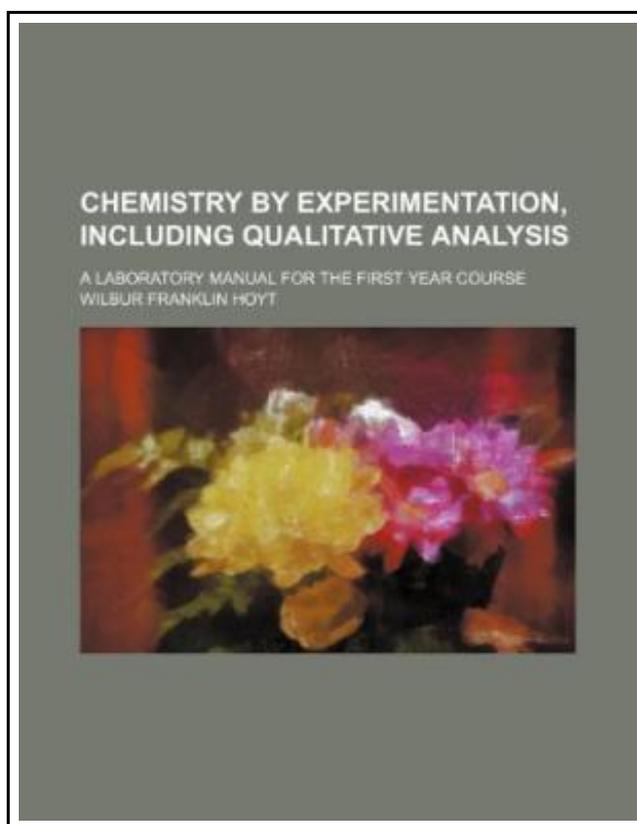


# Chemistry by Experimentation, Including Qualitative Analysis; A Laboratory Manual for the First Year Course



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## ***Reviews***

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*(Aracely Hickle)*

## CHEMISTRY BY EXPERIMENTATION, INCLUDING QUALITATIVE ANALYSIS; A LABORATORY MANUAL FOR THE FIRST YEAR COURSE



To read **Chemistry by Experimentation, Including Qualitative Analysis; A Laboratory Manual for the First Year Course** eBook, make sure you access the link below and download the ebook or get access to additional information which might be in conjunction with CHEMISTRY BY EXPERIMENTATION, INCLUDING QUALITATIVE ANALYSIS; A LABORATORY MANUAL FOR THE FIRST YEAR COURSE ebook.

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1917 Excerpt: .as well as possible with a glass plate until combustion ceases. (E). Note odor. Eemove combustion cup and lower into the bottle a colored flower, or moist colored cloth. (?). After a few minutes, remove cloth or flower, cover the mouth of the bottle with the hand and shake vigorously. Observe action on the litmus solution. (?). What action did the fumes have on flower and cloth? Mention some other bleaching agents. 7. Boil a little S in a t.t. and drop into the boiling liquid a strip of Cu foil. Is there evidence of chemical change? What is formed and how? (E). Add a little HCl to the compound, heat gently and note odor. (?). (E). QUESTIONS AND SUGGESTIONS 1. Write the formulae for the sulfid, hyposulfite, sulfite, sulfate, meta (or per) sulfate, thiosulfate, and pyrosulfate of sodium. Is there any variance in naming any of the above? 2. Mention several uses of S and its compounds. 3. What tests for S. have we had in this exercise? THE HALOGENS Materials: Solid; NaCl, Sb powder, I, MnO<sub>2</sub>, plant or colored cloth. Solution; NaI, KI, KBr, AgNO<sub>3</sub>, . H<sub>2</sub>SO<sub>4</sub>, and NH<sub>4</sub>OH. 1. Recall II, 3; VII, 5 (d); XV, 6, and restate purpose and results. Compare the solubility of a crystal or two of I in 3 c.c. of water, and in 3 c.c. KI. (S--for5.) 2. To 5 c.c. solutions of NaCl, KI, and KBr in separate t.t.s, add a little strong H<sub>2</sub>SO<sub>4</sub> and heat...

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